Formation of Novel Photoadducts from Irradiation of 5(S)-5-O-tert-Butyldimethylsiloxymethylfuran-2(5H)-one

David Brown, Christine J. Cardin and John Mann*

Department of Chemistry, Reading University, Whiteknights, Reading, UK RG6 2AD

Irradiation of 5*S*-5-*O*-tert-butyldimethylsiloxymethylfuran-2(5*H*)-one **1** in acetonitrile yields the *C*2-symmetric bis(lactone), 1*S*,2*S*,6*S*,7*S*-[3*S*,10*S*]-bis-tert-butyldimethylsiloxymethyl-4,9-dioxatricyclo[5.3.0.0^{2,6}]deca-5,8-dione **6**, and a 3-substituted intramolecular product resulting from an apparent 8-endo-trig cyclisation.

We have been exploring the photocatalysed additions of alcohols¹ and amines² to stereochemically defined 5-substituted-furan-2(5*H*)-ones, *e.g.* **1**, and had demonstrated that ethylene glycol would add under irradiation (254 nm) to yield a mixture of diastereoisomers **2a** and **2b** [(4*R*,5*S*)-5-*tert*-butyldimethylsiloxymethyl-4-(*R*,*S*)-dihydroxyethyltetra-hydrofuran-2(5*H*)-one] which could be separated after conversion into the respective acetonides. Alternatively, irradiating **1** in the presence of 2,2-dimethyl-1,3-dioxolane yields the adducts **3a** and **3b**[†] {(4*R*,5*S*)-5-*tert*-butyldimethylsiloxymethyl-4-[*O*-isopropylidene-(*R*,*S*)-dioxyethyl]tetrahydrofuran-2(5*H*)-one} directly (13, 17% respective isolated yields after chromatography).

Variable quantities of other products were also obtained, and we initially believed that these might be further addition products between adducts 3 and butenolide 1. In order to try to improve the synthesis of these adducts, butenolide 1 was irradiated (MeCN, 254 nm) in the presence of an excess of the (S)-isomer 3a, and after 16 h two new products were formed. From the NMR data, one of these was clearly a butenolide dimer, and the other an intramolecular photoaddition product, which after close inspection of the NMR data was assigned the structure 4.[‡] Such an addition is, to our knowledge, the first example of a photochemically-induced intramolecular addition to 5-substituted-furan-2(5H)-ones, and of a photochemical addition that occurs at the 3 (rather than 4) position. The reaction presumably occurs by the mechanism shown in Scheme 1. The initial 1,3-biradical abstracts a hydrogen from the tert-butyl group and the resultant primary radical, which is stabilised by the β -silicon atom, reacts with the incipient radical at C-3 to form compound 4. Subsequent hydrolysis (aqueous AcOH in THF) provides compound 5.§



The dimer was more efficiently prepared by irradiation of butenolide 1 alone (0.06 mol dm⁻³ solution in MeCN, 254 nm, 16 h). This resulted in the formation of major **6**¶ and minor **7**∥ dimeric products (24, 12% respective isolated yields after chromatography), and the intramolecular photoadduct **4** (9%). As the major dimer **6** contained magnetically equivalent monomeric units, the structure was determined by X-ray crystallography (Fig. 1) and shown to be the head-to-head (HH), *cis-anti-cis* isomer.** Although residual disorder in the TBDMS sidechain containing Si(13) precluded refinement to a lower *R* factor, the stereochemistry of the cyclobutane ring was clearly established.

Despite extensive NMR studies, signal overlap has so far precluded structure determination of the minor dimeric product 7. Work is being carried out to improve the yield of the C2-symmetric bis(lactone), which has considerable synthetic





Fig. 1 X-Ray structure of 6

potential, and to investigate the potential of the intramolecular photocyclisation reacton, which to our knowledge is unprecedented.

D. B. thanks the EPSRC for a studentship.

Received, 9th January 1995; Com. 5/00151J

Footnotes

† The two anomers were separable by flash chromatography (diethyl etherlight petroleum, 3:2) to give two eluted anomers. First eluted anomer isolated as a white crystalline solid: mp 55–56 °C; $\delta_{\rm H}$ (CDCl₃, 250 MHz; *J*/Hz) 0.07 (3H, s, SiCH₃), 0.08 (3H, s, SiCH₃), 0.89 [9H, s, C(CH₃)₃], 1.35 (3H, s, acetonide CH₃), 1.43 (3H, s, acetonide CH₃), 2.52 (1H, dd, *J*_{gem} 20.7, *J*_{3,4} 8.9, H-3a), 2.61–2.73 (2H, m, H-3b, H-4), 3.59 (1H, dd, *J*_{gem} 8.3, *J*_{2',1'} 6.4, H-2a'), 3.70 (1H, dd, *J*_{gem} 11.2, *J*_{6,5} 2.8, H-6a), 3.87 (1H, dd, *J*_{gem} 11.3, *J*_{6,5} 3.4, H-6b), 4.08 (1H, dd, *J*_{gem} 8.3, *J*_{2',1'} 6.5, H-2b'), 4.18–4.25 (1H, m, H-1'), 4.37–4.40 (1H, m, H-5). Second eluted anomer isolated as a colourless oil: $\delta_{\rm H}$ (CDCl₃, 250 MHz) 0.06 (3H, s, SiCH₃), 0.07 (3H, s, SiCH₃), 0.89 [9H, s, C(CH₃)₃], 1.34 (3H, s, acetonide CH₃), 2.19 [1H, dd, *J*_{gem} 20.7, *J*_{3,4} 8.9, H-3a), 2.64–2.66 (1H, m, H-4), 2.74 (1H, dd, *J*_{gem} 16.9, *J*_{3,4} 9.8, H-3b), 3.62–3.65 (1H, m, H-2'a), 3.69 (1H, dd, *J*_{gem} 11.3, *J*_{6,5} 2.6, H-6a), 3.93 (1H, dd, *J*_{gem} 11.3, *J*_{6,5} 2.8, H-6b), 4.05 (1H, m, H-2'a), 4.05 (1H, m, H-2'a), 3.69 (1H, dd, *J*_{gem} 11.3, *J*_{6,5} 2.6, H-6a), 3.93 (1H, dd, *J*_{gem} 11.3, *J*_{6,5} 2.8, H-6b), 4.05–4.13 (2H, m, H-1', H-2b'), 4.54–4.58 (1H, m, H-5).

‡ 4,4,5,5-Tetramethyl-3,9-dioxa-4-sila-bicyclo[5.2.1]decan-8-one 4 isolated as a white crystalline solid: mp 87–88 °C; ν_{max} (Nujol mull)/cm⁻¹ 2923vs, 2853vs, 1757vs, 1342s, 1185s, 1154s; $\delta_{\rm H}$ (CDCl₃, 400 MHz) 0.11 (3H, s, SiCH₃), 0.29 (3H, s, SiCH₃), 0.84 (3H, s, CCH₃), 1.09 (3H, s, CCH₃), 1.82 (1H, dd, $J_{\rm gem}$ 15, $J_{6,7}$ 3.7, H-6a), 1.97–2.05 (2H, m, H-6b), H-10a), 2.65–2.74 (1H, m, H-10b), 2.84–2.91 (1H, m, H-7b), 3.74 (1H, dd, $J_{\rm gem}$ 12.5, $J_{1,2}$ 0.7, H-2a), 4.19 (1H, dd, $J_{\rm gem}$ 12.5, $J_{1,2}$ 2.6, H-2b), 4.53–4.57 (1H, m, H-1); *m/z* (GC-MS/EI) 229 (MH⁺).

(3R,5S)-Hydroxymethyl-3-[2'-dimethylsilylhydroxy-2',2'-dimethyl-ethyl]tetrahydrofuran-2(5H)-one **5** isolated as a white crystalline solid: mp 97–98 °C; v_{max} (Nujol mull)/cm⁻¹ 3306s, br, 3220s, br, 2920vs, 2852vs, 1753vs, 1296w, 1254s, 1203s, 1176w, 1112w, 1057s, 866s, 829w; $\delta_{\rm H}$ (CDCl₃, 400 MHz) 0.13 (3H, s, SiCH₃), 0.15 (3H, s, SiCH₃), 0.93 (3H, s, CCH₃), 0.96 (3H, s, CCH₃), 1.34 (1H, dd, $J_{\rm gem}$ 14, $J_{1'a,3}$ 8.6, H-1'a), 1.79 [1H, br s, Si(CH₃)₂OH], 1.82–1.88 (1H, m, H-4a), 2.2 (1H, dd, $J_{\rm gem}$ 14, $J_{1'rb,3}$ 2.8, H-1'b), 2.33 (1H, br s, CH₂OH), 2.46–2.53 (1H, m, H-4b), 2.77–2.85 (1H, m, H-3), 3.62 (1H, dd, $J_{\rm gem}$ 12, $J_{6,5}$ 4.5, H-6a), 3.62 (1H, dd, $J_{\rm gem}$ 12, H-6b), 4.46–4.52 (1H, m, H-5).

 $\begin{array}{ll} 1S,2S,6S,7S-[3S,10S]-Bis-tert-butyldimethylsiloxymethyl-4,9-diox$ atricyclo[5.3.0.0^{2.6}]deca-5,8-dione**6**isolated as a white crystalline solid:mp 180–181 °C; Satisfactory elemental analysis obtained; v_{max} (Nujol $mull)/cm⁻¹ 2924vs, 2852vs, 1774s, 1758s, 1337s, 1252s, 1161s, 834s; <math>\delta_{\rm H}$ (CDCl₃, 400 MHz) 0.04 (6H, s, SiCH₃ × 2), 0.05 (6H, s, SiCH₃ × 2), 0.85 [18H, s, (CH₃)₃ × 2], 3.06 (2H, d, J 5.5), 3.22 (2H, d, J 5.5), 3.65 (2H, dd, J_{gem} 1.2, J 1.8), 3.85 (2H, dd, J_{gem} 11.2, J 2.7), 4.54 (2H, m, H-3, H-10); m/z (GC-MS/EI) 399 [M⁺ – C(CH₃)₃].

 $\begin{array}{l} \left\|\begin{array}{l} 5S,10S\text{-Bis-tert-butyldimethylsiloxymethyl-4,9-dioxatricyclo} [5.3.0.0^{2.6}] \\ \text{deca-3,8-dione or } 3S,10S\text{-bis-tert-butyldimethylsiloxymethyl-4,9-dioxatricyclo} [9.5,10] \\ \text{deca-3,8-dione or } 3S,10S\text{-bis-tert-butyldimethylsilox} [9.5,10] \\ \text{deca-3,8-dione or } 3S,10S\text{-bis-tert-butyldimethyls} [9.5,10] \\ \text{deca-3,8-dione or } 3S,10S\text{-bis-4,1} \\ \text{deca-3,8-dione o$

Crystal data for 6, 1S,2S,6S,7S-[3S,10S]-bis-tert-butyldimethylsiloxymethyl-4,9-dioxatricyclo[5.3.0.0^{2,6}]deca-5,8-dione: monoclinic. space group $P2_1$, a = 6.555(8), b = 11.619(8), c = 17.612(7) Å, $\beta =$ $98.61(2)^\circ$, U = 1326.3(2) Å³, Z = 2. The data was collected on a Marresearch Image Plate system, 95 frames at 2° intervals were measured for 2 min each. 3738 Reflections were measured, of which 3629 were independent in a monoclinic unit cell ($R_{int} = 0.0404$), Cell dimensions were determined from analysis of 492 independent reflections. The structure was solved by direct methods using SHELX86 (G. M. Sheldrick, SHELX86, Acta Crystallogr., Sect. A, 1990, 46, 467). The structure was refined on F² using SHELXL (G. M. Sheldrick, SHELXL, program for crystal structure refinement, University of Göttingen). The conventional R-factor for the 1711 reflections with $I > 2\sigma(I)$ was 0.0865. Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Information for Authors, Issue No. 1.

References

- 1 J. Mann and A. Weymouth-Wilson, Synlett, 1992, 67; J. Chem. Soc., Perkin Trans. 1, 1994, 3141.
- 2 E. Santiago de Alvarenga and J. Mann, J. Chem. Soc., Perkin Trans. 1, 1993, 2141.